

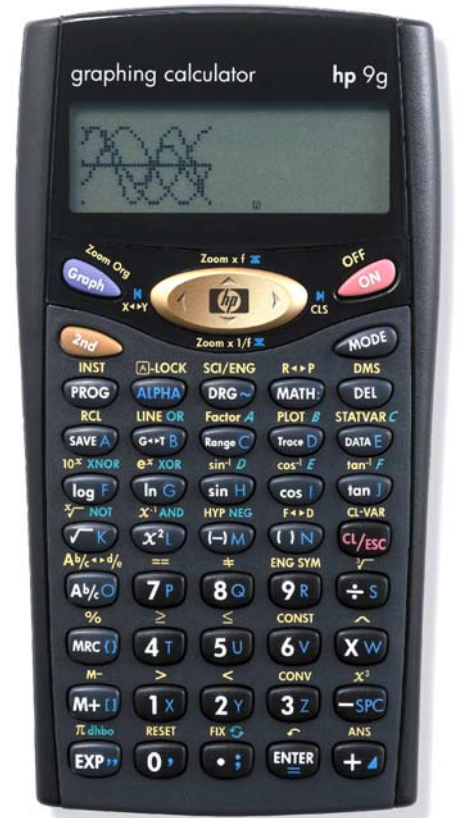


hp calculators

HP 9g Using the Built-in Physical Constants

The CONST Menu

Practice Solving Problems Involving Physical Constants



The CONST menu

The HP 9g includes twenty physics constants that can be used in calculations and included in your programs. All the constant are expressed in SI units, but only the values are present. For example, for the speed of light, the built-in value is 299792458 which is in m/s, even though units are not stated.

For your convenience, the following table lists all the constant in the order they appear in the CONST menu, along with the keystroke sequence to retrieve their values in the entry line:

Symbol	Meaning	Value	Keys
c	Speed of light	299792458	$\text{2nd} \text{CONST} \text{0}$
Vm	Molar volume of ideal gas	0.0224141	$\text{2nd} \text{CONST} \text{1 X}$
Na	Avogadro's number	6.022136736e23	$\text{2nd} \text{CONST} \text{2 Y}$
g	Acceleration of gravity	9.80665	$\text{2nd} \text{CONST} \text{3 Z}$
me	Electron mass	9.109389754e-31	$\text{2nd} \text{CONST} \text{4 T}$
G	Gravitational constant	6.6725985e-11	$\text{2nd} \text{CONST} \text{5 U}$
mP	Proton mass	1.67262311e-27	$\text{2nd} \text{CONST} \text{6 V}$
e	Elementary charge	1.602177335e-19	$\text{2nd} \text{CONST} \text{2nd} \text{CONST} \text{0}$
mn	Neutron constant	6.62607554e-34	$\text{2nd} \text{CONST} \text{2nd} \text{CONST} \text{1 X}$
ϵ_0	Dielectric permittivity	8.854187818e-12	$\text{2nd} \text{CONST} \text{2nd} \text{CONST} \text{2 Y}$
h	Planck's constant	6.62607554e-34	$\text{2nd} \text{CONST} \text{2nd} \text{CONST} \text{3 Z}$
φ_0	Flux quantum	2.067834616e-15	$\text{2nd} \text{CONST} \text{2nd} \text{CONST} \text{4 T}$
k	Boltzmann's constant	1.38065812e-23	$\text{2nd} \text{CONST} \text{2nd} \text{CONST} \text{5 U}$
a ₀	Bohr radius	5.291772492e-11	$\text{2nd} \text{CONST} \text{2nd} \text{CONST} \text{6 V}$
R	Gas constant	8.3145107	$\text{2nd} \text{CONST} \text{^}$ 0
μ_0	Magnetic permeability	0.000001257	$\text{2nd} \text{CONST} \text{^}$ 1 X
μ_B	Bohr magneton	9.274015431e-24	$\text{2nd} \text{CONST} \text{^}$ 2 Y
F	Faraday constant	96485.30929	$\text{2nd} \text{CONST} \text{^}$ 3 Z
μ_N	Neutron magnetic moment	5.050786617e-27	$\text{2nd} \text{CONST} \text{^}$ 4 T
μ	Atomic mass constant	1.66054021e-27	$\text{2nd} \text{CONST} \text{^}$ 5 U

The CONST menu is displayed by $\text{2nd} \text{CONST}$. It consists of three menus that display six or five constant each. You can use the arrow keys to scroll through the menus and select the desired constant. Once selected press enter to insert that constant in the entry line, at the current position. Notice that it is the symbol that is inserted in the entry line, not the value. Also the shortest keystroke sequences are shown in the above table: note that when the CONST menu is displayed, pressing $\text{2nd} \text{CONST}$ displays the next menu of constants.

Constants have to entered using the CONST menu, and cannot be typed in ALPHA mode. In fact, some constants that could be mistaken for variable names (e.g. R, F, G) are displayed bold for this purpose.

The HP 9g constants are not available in the BaseN mode.

Practice solving problems involving physical constantsExample 1: Find Plank's constant and view its full value.Solution: Plank's constant, named "h", is in the second submenu of CONST, so the shortest way to *view* this value is pressing the keys:

If you want to invoke this constant in the entry line, press now that it is selected. To view Plank's constant in the result line press again.

Answer: 6.622607554x10⁻³⁴Example 2: Find the quotient between the proton mass and the electron mass.Solution: This ratio is one of the built-in constants on the HP48GX and the HP49G+ and is called "mpme." Although it's not in the above list, you can calculate its value by pressing:Answer: mP/me= 1836.152756. Compare it with the value on the HP 48GX (1836.152701) and the one calculated on the HP 33S (1836.1526633). More precision in measures imply new standard values. Constants are not that constant after all!Example 3: Find the flux density at 5 meters from a long, straight wire that carries current of 9 A.Solution: The field due to the wire is of magnitude:

$$B = \frac{\mu_0 I}{2\pi r}$$

where r is the perpendicular distance from wire. In our case, r = 5 and I = 9. μ_0 is the magnetic permeability of free space, its value is $4\pi \cdot 10^{-7} \text{ Hm}^{-1}$ and is one of the constants in your HP 9g. To find the flux density press:

Note the implicit multiplication after the constant name.

Answer: 0.00000036 TExample 4: Einstein observed that any change in mass implies a change in energy and vice versa. Calculate the energy change corresponding to a change in mass of 1Kg.Solution: Einstein's most famous equation states that:

$$\Delta E = \Delta m \cdot c^2$$

c is the speed of the light in the free space, and we can use the value built in our calculator. Since the change of mass is 1 in our example, the problem is reduced to square c. Press:

2nd CONST 0 x^2L ENTER

Answer: 8.987551787 · 10¹⁶ J

Example 5: A cylindrical tank contains 4 kg of carbon monoxide at -50°C (223K). The tank is 1 meter in length and its inner diameter is 0.2 m. Calculate the pressure of the gas using Van de Waals' equation.

Solution: This equation explains derivations from ideal gas behavior and is:

$$\left(P + \frac{a}{V^2}\right)(V - b) = RT$$

where a and b are constants characteristics of a particular gas; for the carbon monoxide they are 0.1474 and 3.95 · 10⁻⁵ respectively. Solving for P:

$$P = \frac{RT}{V - b} - \frac{a}{V^2}$$

Before starting to calculate, notice that V is actually the molar volume and, when expressed in m³, is equal to:

$$V = \frac{7}{4000} \pi \cdot 0.2^2$$

Since it occurs twice in the pressure equation, we'll calculate this value first and use the ANS function to retrieve V. Press:

\cdot 2Y x^2L 2nd π dbb 7P \div 4T EXP 3Z ENTER

and now we can calculate P by pressing:

2nd CONST \wedge 0 2Y 2Y 3Z \div ANS 2nd ANS ←SPC 3Z \cdot 9R 5U EXP ↔M 5U ←SPC
 \cdot 1X 4T 7P 4T \div 2nd ANS x^2L ENTER

Answer: 7229354.15 Pa or 72.3 bar.

Example 6: Find the revolution period of an artificial satellite put into circular orbit around the Earth. The radius of the orbit is 8000 km.

Solution: The period is given by:

$$T = 2\pi\sqrt{\frac{R^3}{GM}}$$

where R is the radius of the orbit, M is the mass of the Earth ($5.98 \cdot 10^{24}$ kg) and G is the gravitational constant. (Notice that it does *not* depend on the mass of the satellite). The period can be calculated by pressing:



Answer: 7117.324386 s or 2 hours approximately.